

# Structural and energetic determinants of tyrosylprotein sulfotransferase sulfation specificity

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## ABSTRACT

**Motivation:** Tyrosine sulfation is a type of post-translational modification (PTM) catalyzed by tyrosylprotein sulfotransferases (TPST). The modification plays a crucial role in mediating protein-protein interactions in many biologically important processes. There is no well-defined sequence motif for TPST sulfation and the underlying determinants of TPST sulfation specificity remains elusive. Here, we perform molecular modeling to uncover the structural and energetic determinants of TPST sulfation specificity.

**Results:** We estimate the binding affinities between TPST and peptides around tyrosines of both sulfated and nonsulfated proteins in order to differentiate them. We find that better differentiation is achieved after including energy costs associated with local unfolding of the tyrosine-containing peptide in a host protein, which depends on both the peptide's secondary structures and solvent accessibility. Local unfolding renders buried peptide with ordered structures thermodynamically available for TPST binding. Our results suggest that both thermodynamic availability of the peptide and its binding affinity to the enzyme are important for TPST sulfation specificity, and their interplay results into great variations in sequences and structures of sulfated peptides. We expect our method useful in predicting potential sulfation sites and transferable to other TPST variants. Our study may also shed light on other PTM systems without well-defined sequence and structural specificities.

**Availability and implementation:** All the data and scripts used in the work are available at <http://dlab.clemson.edu/research/Sulfation>

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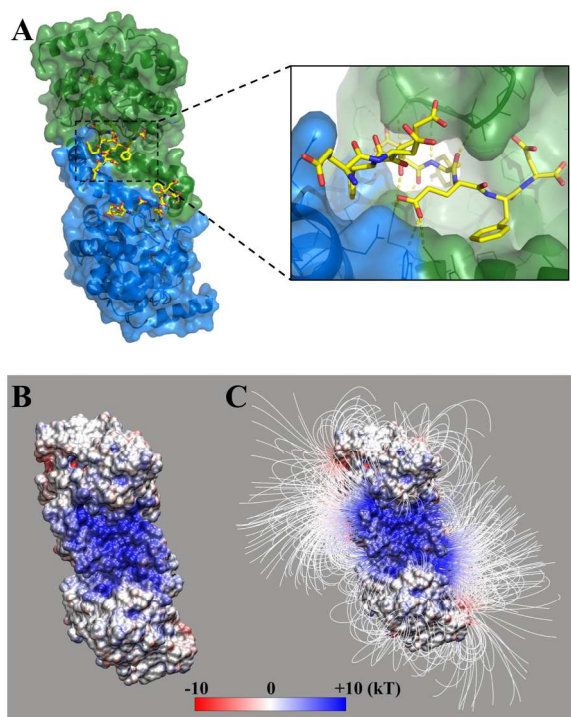
**Supplementary Information:** Supplementary data are available at the journal's web site.

## 1 INTRODUCTION

After their synthesis in the ribosome, many proteins undergo post-translational modifications (PTM) such as glycosylation, phosphorylation, and peptide hydrolysis before reaching their fully functional forms. Tyrosine sulfation is a common PTM occurring on many proteins that transit through the Golgi apparatus, such as extracellular matrix proteins, serine protease inhibitors, and G-protein coupled receptors (Stone *et al.*, 2009). So far hundreds of tyrosine-sulfated proteins have been identified, and more are likely to be discovered. A list of sulfated proteins can be obtained from the UniProt database (Bairoch *et al.*, 2005). A major functional

role of tyrosine sulfation is to mediate protein-protein interactions (Kehoe and Bertozzi, 2000). Sulfation is vital for many biological functions as indicated in studies showing that in the absence of sulfation, postnatal viability, vision, fertility and growth are affected in mice (Ouyang *et al.*, 2002; Borghei *et al.*, 2006). The function of many proteins, including P-selectin glycoprotein ligand-1 (Pouyani and Seed, 1995; Wilkins *et al.*, 1995), chemokine receptors (Simpson *et al.*, 2009), platelet glycoprotein Ib (Zarpellon *et al.*, 2011; Uff *et al.*, 2002), depend on tyrosine sulfation. For example, the sulfation of tyrosine in the chemokine receptor CCR5 is necessary for HIV-1 gp120 mediated entry of HIV into CD4+ T-lymphocytes (Tyrosine Sulfation of Human Antibodies Contributes to Recognition of the CCR5 Binding Region of HIV-1 gp120, 2003). A detailed understanding of the molecular mechanism of tyrosine-sulfation is therefore important for manipulating such modifications, regulating cell signaling, and drug development.

Sulfation is catalyzed by the tyrosylprotein sulfotransferase (TPST) enzymes, which reside inside the Golgi apparatus. The process involves the transfer of a sulfo group from a bound 3'-phosphoadenosine-5'-phosphosulfate (PAPS) to the phenol group of tyrosine. In humans, two isoforms, TPST-1 and TPST-2, are found with 64% sequence identity between them (Teramoto *et al.*, 2013). The functional differences between the isoforms or the necessity of two such isoforms are not well established. Sulfation occurs only on specific tyrosines in proteins. Even though up to 1% of tyrosines in a cell's proteome can be sulfated (Önnerfjord *et al.*, 2004), the location of sulfated tyrosines are not known for many proteins. Hence, neither the exact role of sulfated tyrosines in these proteins nor the mechanism by which the tyrosines are selected for sulfation is fully understood. Analysis of the amino acid sequences flanking sulfated tyrosines suggested some general features such as the presence of acidic and small residues, absence of disulfide bonds or glycosylated residues, and a reduced number of hydrophobic residues in the vicinity of sulfated tyrosines (Rosenquist and Nicholas, 1993). However, there are also many exceptions to these general characteristics of sequence specificity. For example, mutational studies reported an enhanced sulfation efficiency when tyrosines were flanked by basic residues (Bundgaard *et al.*, 1997). Therefore, in contrast to other PTMs such as N-glycosylation, which recognizes structurally available NX(T/S) triplets (Marshall, 1974), a well-defined sequence specificity for tyrosine sulfation cannot be established.



**Figure 1. The structure of TPST-2 dimer.** (A) The TPST-2 dimer (Teramoto *et al.*, 2013) with bound substrate peptide (yellow) and PAPS analog. (B) Electrostatic potential and (C) field lines of the enzyme after removing the substrate peptide. The binding pocket has a high positive electrostatic potential, and hence the peptides with net negative charge can be driven into the pocket by electrostatic interactions

Due to the lack of obvious sequence patterns, sophisticated statistical tools have been developed to predict potential locations of sulfation sites in a given protein. Sulfinator (Monigatti *et al.*, 2002) constructed four different Hidden Markov Models to recognize sulfated tyrosine residues depending on their locations in the sequence: near N-terminal, near C-terminal, in the center of a window of at least 25 amino acids, and in windows containing multiple tyrosines. In PredSulSite (Huang *et al.*, 2012), physicochemical properties of amino acids along with predicted secondary structures and amino acid sequence order are considered in a supported vector machine (SVM). The SVM algorithm has also been applied in another predictor based on predicted secondary structures and solvent accessible surface area (Chang *et al.*, 2009). These statistics-based tools work satisfactorily in their test cases, and have been useful in experimental studies of protein sulfation (Goff *et al.*, 2003; Keykhosravani *et al.*, 2005). However, these training-based statistical methods depend heavily on the quality of the training set such as the coverage and completeness, which is not the case for the increasing list of sulfated proteins identified experimentally. As a result, many exceptions were observed in cases beyond the training sets (Önnerfjord *et al.*, 2004), (Monigatti *et al.*, 2006), and these methods are also not transferable to other TPST variants, such as recently discovered bacterial TPSTs (Han *et al.*, 2012). Most importantly, many of these methods are sequence-based and the constructed predictors lack structural or physicochemical insights to the molecular mechanism of tyrosine selection by TPST enzyme.

Recently, a high-resolution structure of human TPST-2 (Fig. 1A) has been solved at 1.9 Å resolution in complex with a high

affinity peptide using X-ray crystallography (Teramoto *et al.*, 2013), making it possible to study the structural and energetic determinants of TPST sulfation specificity. In this work, we adapt Eris (Yin *et al.*, 2007a), a method developed to compute protein stability changes upon mutations, to estimate the binding affinities between TPST-2 and various peptide substrates in order to differentiate sulfated and nonsulfated sequences that have been experimentally verified. We find that the peptide-TPST binding affinities cannot separate the sulfated and nonsulfated sequences satisfactorily. Better differentiation is achieved after including energy costs associated with local unfolding of the tyrosine-containing peptides in the host protein, which depends on both the peptide's secondary structures and solvent accessibility. The thermodynamic population of the locally unfolded peptides determines the availability of the peptide for TPST binding and subsequent catalysis. Therefore, our study suggests that both the thermodynamics accessibility of a peptide and its binding affinity to TPST are important for sulfation. The interplay of these two factors allow a great variety in sequences and structures of sulfated peptides, where a buried peptide with well-defined secondary structure might be sulfated if the peptide undergoes local unfolding, making itself available for enzyme binding.

## 2 METHODS

**Electrostatic analysis.** In order to determine how the electrostatic interactions guide the peptides to the protein, the electrostatic potential and force surrounding the protein was calculated using Delphi (Li *et al.*, 2012) after removing the peptide from the binding site. The following parameters were used for the calculation: scale=2.0 grids/Å; grid size 280×280×280; and dielectric constants of 2.0 and 80.0 for protein and water environment, respectively. The force field used for the calculations was AMBER (Lindorff-Larsen *et al.*, 2010). VMD (Humphrey *et al.*, 1996) was used to visualize the electrostatic surface and field lines.

**Protein-peptide binding affinity.** The relative binding affinity of a given peptide with respect to the reference peptide can be quantified as,  $\Delta\Delta G_{bind} = (G^{complex} - G^{prot} - G^{pep})_{mut} - (G^{complex} - G^{prot} - G^{pep})_{ref}$ , where the superscript *complex* denotes the enzyme-peptide complex; *prot* and *pep* refer to the protein and peptide in their unbound states, respectively; the subscript *mut* denotes mutations of a given peptide with respect to the reference peptide indicated by the subscript *ref*. Because the absolute free energy is difficult to measure, the free energy difference between the folded and unfolded states, i.e. the stability  $\Delta G$ , is most commonly used. The unfolded state of the complex corresponds to the unfolded protein and peptide, and thus  $\Delta\Delta G_{bind} = (\Delta G^{complex} - \Delta G^{prot} - \Delta G^{pep})_{mut} - (\Delta G^{complex} - \Delta G^{prot} - \Delta G^{pep})_{ref}$ . Since the protein sequence is not changing,

$$\Delta\Delta G_{bind} = \Delta\Delta G^{complex} - \Delta\Delta G^{pep}, \quad \text{Eq. (1)}$$

where  $\Delta\Delta G^{complex}$  and  $\Delta\Delta G^{pep}$  refer to the mutation-induced stability changes for the complex and peptide, respectively.

**Estimation of stability change upon mutations.** We used Eris to estimate the stability changes upon mutations (Yin *et al.*, 2007a). Eris uses the backbone-dependent rotamer library to model protein side-chain conformations (Dunbrack and Cohen, 1997). Given the vast side-chain rotameric space, the optimal packing was searched via a Monte-Carlo based simulated annealing algorithm, where the rotameric space of side-chains was sampled according to the Metropolis criteria and the simulation temperature is gradually reduced till the acceptance rate is below a pre-defined threshold. The stability of a given sequence and corresponding structural conformation was evaluated with the Medusa force field (Ding and Dokholyan, 2006) which includes van der Waals, solvation, hydrogen bonds, electrostatics, statistical potential for backbone-dependent amino acid identify and rotamer, and reference energy for the unfolded states. The atom types and corresponding van der Waals interaction parameters were taken from

CHARMM (Brooks *et al.*, 1983). The solvation energy was approximated by the Lazaridis-Karplus implicit solvent model (Lazaridis and Karplus, 1999). The distance- and angle-dependent hydrogen bond interaction parameters were adapted from ref. (Kortemme and Baker, 2002). We used the Debye-Hückel approximation to model the screened charge-charge interactions at the physiological condition, namely pH  $\sim 7$  and salt concentration  $\sim 0.1$  M, and the corresponding Debye length was approximately 10 Å. The weights for different energy terms and the reference energy of unfolded state were determined by recapitulating the native amino acid sequence of a set of high-resolution protein structures (Yin *et al.*, 2008).

Due to the stochastic nature of the simulated annealing algorithm, multiple simulations were often performed in order to compute the average stability of a given sequence. We used the PDB structure of 3AP1 (Teramoto *et al.*, 2013) to model the bound complexes (Fig. 2A). We chose a 9-residue window with the sulfated tyrosine as the sixth residue similar to the reference peptide in the crystallography structure. It has been experimentally shown that these flanking residues are important for both TPST binding and catalysis (Lin *et al.*, 1992). For a given mutation, 100 independent simulations were performed for both the native and mutant protein (or protein-peptide complex). The stability was estimated as the average value over all simulations, and the mutation-induced stability change was then obtained as the difference between mutant and wild type.

**Positive and negative data sets.** In order to test whether a parameter can be used to differentiate the sulfated and nonsulfated proteins, we compiled a list of experimentally verified sulfated peptide/proteins (positive dataset) and a list of nonsulfated peptide/proteins (negative dataset). For the positive dataset, we collected a list of 160 non-redundant tyrosine-sulfated peptide/proteins by combining both Sulfinator (Monigatti *et al.*, 2002) and dbPTM (Lu *et al.*, 2012) datasets, which were extracted from the UniProt database (Bairoch *et al.*, 2005). Many proteins had more than one sulfated tyrosine. Similarly, based on Sulfinator, a list of 159 peptide-proteins was constructed for the secreted proteins that do not undergo sulfation.

**Secondary structures and relative solvent accessibility predictions.** We used the NetSurfP (Petersen *et al.*, 2009), a protein surface accessibility and secondary structure prediction web-server, to estimate a peptide's structural propensity in terms of relative solvent accessibility,  $P_{rsa}$ , and various secondary structures,  $P_{\alpha}$ ,  $P_{\beta}$ , and  $P_{coil}$ . A benchmark study by the developers (Petersen *et al.*, 2009) indicated that the prediction accuracy of NetSurfP is comparable to other best publicly available methods. With the input of the sequence of a host protein, the profiles of relative solvent accessibility and secondary structures –  $\alpha$ -helix,  $\beta$ -sheet, and random coil – were obtained for all residues. We computed the average propensities,  $P_{\alpha}$ ,  $P_{\beta}$ ,  $P_{coil}$ , and  $P_{rsa}$  over the 9-residue window around a given tyrosine of interest.

**Z-score and Z-score minimization.** For a given parameter,  $\chi$ , derived for both the sulfated and nonsulfated sequences, we defined the Z-score:

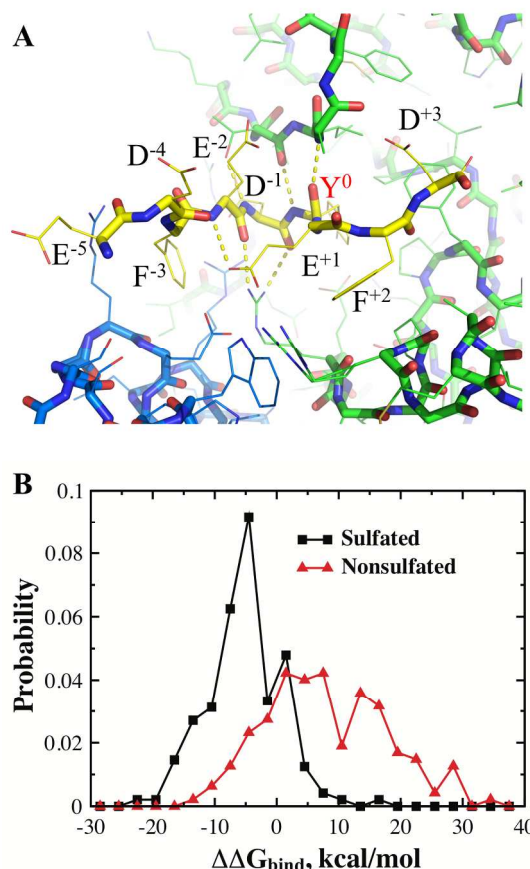
$$Z = \frac{\langle \chi \rangle_{sulfated} - \langle \chi \rangle_{nonsulfated}}{\delta(\chi)_{sulfated} + \delta(\chi)_{nonsulfated}}, \quad \text{Eq. (2)}$$

where  $\delta\chi$  is the standard deviation. The averaging was done separately for all peptides belonging to sulfated and nonsulfated datasets. The Z-score quantifies the separation between the  $\chi$ -value distributions of the two datasets.

In order to determine the coefficients  $E_{\alpha}$ ,  $E_{\beta}$ , and  $E_{rsa}$  in the effective energy  $E^{eff}$ , we minimized the Z-score of  $E^{eff}$  using a Monte-Carlo-based simulated annealing. We gradually decreased the Monte-Carlo temperature, and at each temperature multiple rounds of perturbations of the coefficients were applied. The acceptance or rejection of perturbations was determined according to the Metropolis criteria. As the Monte-Carlo temperature approaches zero, the Z-score was minimized stochastically. Multiple independent simulations with different random seeds were performed to ensure the convergence.

### 3 RESULTS

The TPSP-2 enzyme has an N-terminal cytoplasmic domain, a transmembrane domain anchoring the protein in the Golgi membrane, a putative stem region, and a luminal domain that catalyzes the tyrosine sulfation (Teramoto *et al.*, 2013). The protein exists as a homodimer and the structure of the catalytic domain has been solved recently in its dimeric form (Teramoto *et al.*, 2013) (PDB ID: 3AP1, Fig. 1A). The peptide binds to a well-defined deep pocket, a part of which is at the inter-monomer interface (Fig. 1A).



**Figure 2. The estimation of protein-peptide binding affinity.** (A) The structure of the reference nine-residue peptide in the TPST-2 binding pocket as obtained from x-ray crystallography structure, with the sulfated tyrosine positioned as the sixth residue ( $Y^0$ ). The neighboring residues and their positions (superscript) relative to the tyrosine are also marked. (B) The probability density of peptide-enzyme binding scores ( $\Delta\Delta G_{bind}$ ) for the peptides both sulfated and nonsulfated peptides.

**The role of electrostatics in peptide binding.** The peptide-binding pocket of the enzyme is rich in positively charged residues. Electrostatics analysis using Delphi (Li *et al.*, 2012) (see Materials and Methods) indicates that the electrostatic potential near the binding pocket is highly positive (Fig. 1B). Even more, the electrostatic field lines form an electrostatic funnel, which can drive a negatively charged substrate toward the pocket (Fig. 1C). Indeed, many sulfated peptides have a net negative charge, typically originating from acidic groups positioned near the tyrosine. For example, in many cases the sulfated tyrosines are flanked by two acidic residues, providing local net charge of  $-2e$ . This feature was believed to be a necessary requirement for all sulfated tyrosines (Hortin *et al.*, 1986). The more recent discovery of sulfated tyrosines not flanked by acidic residues has called into question this

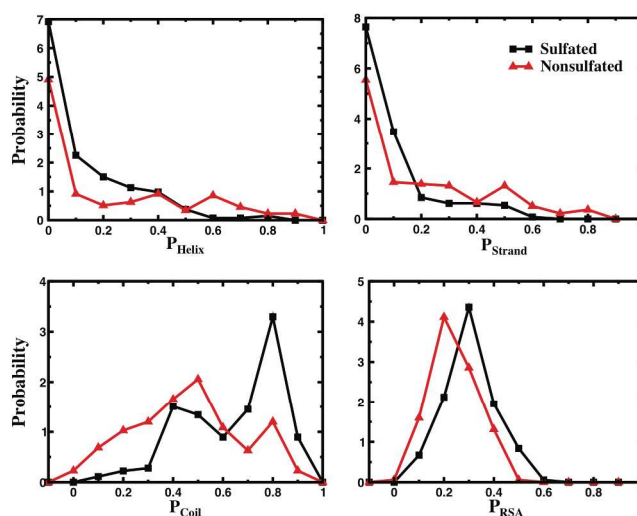


requirement (Bundgaard *et al.*, 1997). However, the net negative charge may still be present due to more distant acidic residues. The observation that TPSP dimer forms such a prominent positive electrostatic patch at the dimer interface suggests that electrostatic interactions between the TPSP and the peptide is an important feature for the binding and perhaps for the specificity of the recognition. To reveal the interplay between binding affinity and specificity, we utilize the protein-peptide complex structure to estimate the binding affinities of both sulfated and nonsulfated peptides in order to investigate the specificity of TPST recognition.

**Protein-peptide binding energy.** In the complex structure, the peptide conformation is stabilized by several hydrogen bonds with the enzyme, including both backbones and side-chains (Fig. 1A) (Teramoto *et al.*, 2013). In order to be catalyzed by the enzyme, a substrate peptide has to assume an appropriate conformation in the pocket such that specific TPST-peptide interactions can be established. Therefore it is expected that the corresponding binding affinity would vary significantly with the amino acid sequences, which in turn would be crucial for the selectivity of peptide sequences for TPST sulfation. Since it is difficult to estimate the absolute value of binding affinity  $\Delta G$ , we computed the change in binding affinity,  $\Delta\Delta G_{bind}$ , due to mutations with respect to the reference peptide in the x-ray crystallography structure of the TPST-peptide complex (see Materials and Methods). We use Eris (Yin *et al.*, 2007a) to estimate the stability changes upon mutations, where the inter-atomic interactions is modeled with the Medusa force field (Ding and Dokholyan, 2006). The Medusa force field includes major physical interactions that govern the binding between peptide and receptor, including van der Waals, solvation, hydrogen bonds and electrostatics (see Materials and Methods). The Eris/Medusa method has been shown efficient in recapitulating protein stability changes upon mutations with a high correlation between predictions and experimental measurements (Yin *et al.*, 2007a, 2007b). For the peptides, a window of nine residues is used with the tyrosine of our interest at the sixth position as for the reference peptide in the complex structure (Fig. 2A).

We first test whether the peptide-binding affinity is the driving force for tyrosine sulfation by TPST. We constructed a list of sulfated and nonsulfated proteins that are experimentally validated (see Materials and Methods). If the peptide-binding energy is the determinant for the TPST sulfation specificity, the sulfated sequences should have stronger affinities or lower binding energies than the nonsulfated sequences. We compute the probability densities of the Eris-derived relative binding affinity  $\Delta\Delta G_{bind}$  for sulfated and nonsulfated sequences (Fig. 2B). As expected, the sulfated sequences, in general, have lower  $\Delta\Delta G_{bind}$  values compared to the nonsulfated sequences. Thus, the peptide-binding affinity plays a crucial role in the sulfation selection process. However, it is also clear from Fig. 2B that a significant separation of the two sets of sequences is not achieved on the basis of  $\Delta\Delta G_{bind}$  values alone. To quantify the separation of two datasets, the standard score (i.e., Z-score, see Materials and Methods) is calculated to determine if the separation of two Gaussian-like distributions is statistically significant. Z-score quantifies the separation with respect to the standard deviations. A larger absolute Z-score value indicates a more significant separation of the two distributions. The Z-score for the two distributions is -0.83, indicating that the separation is within one standard deviation and thus two datasets are not well separated according to the peptide-binding affinity alone. Thus, even though the binding affinity plays an important role in the selection process,

there are additional factors that contribute to the selection of tyrosines by TPST.



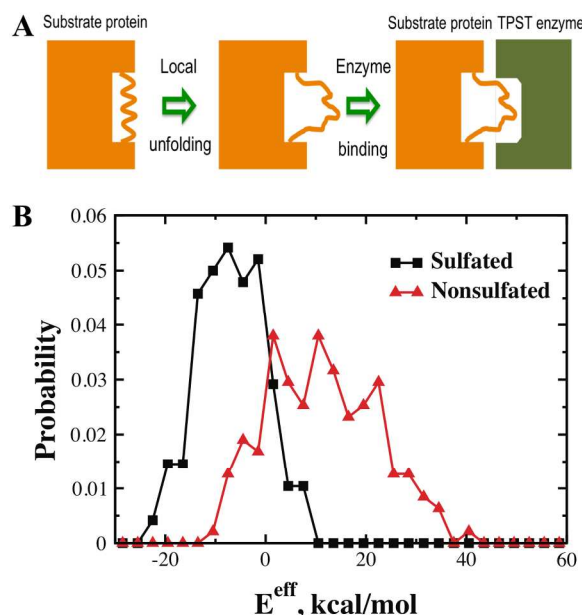
**Figure 3. Secondary structures and solvent accessibility of peptides in host proteins.** The normalized probability density of peptides in the dataset used, with respect to their propensities to form  $\alpha$ -helix,  $\beta$ -strand and random coil secondary structures, and their relative solvent accessibility

**Local unfolding of the tyrosine-containing peptide.** It has been reported that many of the tyrosines that undergo sulfation are positioned in unstructured regions of the host protein, although some sulfated peptides contain ordered secondary structures (Huang *et al.*, 2012), (Chang *et al.*, 2009). In their native structures, peptides that fold into  $\alpha$ -helix or  $\beta$ -sheet structures in the host protein would not be able to fit into the TPST binding pocket (Fig. 1 & 2A). In other words, these peptides with well-defined secondary structures would need to locally unfold in order to bind to the enzyme. The unfolding of secondary structures requires energy, making the binding less favorable. Additionally, in order to bind to the enzyme, the peptide segment also needs to break tertiary contacts, if any, with respect to the rest of the host protein. The energy cost associated with losing these tertiary contacts also makes the recognition by the enzyme unfavorable. The number of tertiary contacts is inversely proportionally to the solvent accessible surface area (Ding *et al.*, 2012). Next, we examine the secondary structures and solvent accessibility of the sulfated and nonsulfated peptides in the corresponding host proteins.

The majority of proteins in the UniProt datasets do not have experimentally determined three-dimension structures available. As a result, the secondary structures and solvent accessibilities of the peptides in our sulfated and nonsulfated datasets cannot be derived directly from the structure of their host proteins. On the other hand, bioinformatics tools have been developed to predict protein secondary structures and solvent accessibility from sequences with significant accuracy. We use the web-server Net-SurfP (Petersen *et al.*, 2009) to estimate the propensity of a peptide in terms of relative solvent accessibility,  $P_{rsa}$ , and secondary structure elements,  $P_{\alpha}$ ,  $P_{\beta}$ , and  $P_{coil}$  (Materials and Methods). We find that compared to the nonsulfated sequences the sulfated ones tend to have weaker propensities for ordered secondary structures (Fig. 3A,B), and consequently higher propensity for random coils (Fig. 3C), although the differences are relative small with major overlaps of the distributions. Similarly, as expected the sulfated

sequences also have slightly higher probability to be solvent exposed than those nonsulfated sequences (Fig. 3D). Therefore, local unfolding of the peptide in the host protein – including both unfolding of the ordered secondary structures and losing tertiary contacts with respect to the rest of the protein, the energy cost of which is inversely proportional to the solvent accessibility – is also important for the recognition of the tyrosine-containing peptide by TPST. The thermodynamic population of the locally unfolded peptides, determined by the energy cost, is available to bind the enzyme. Similar partial unfolding of protein substrates has also been observed for proteolytic cleavage of proteins (Hubbard *et al.*, 1994) as well as for both N- (Marshall, 1974) and O-glycosylation (Hansen *et al.*, 1998) sites.

**An effective energy for sulfation.** Our analysis above suggests that the thermodynamic availability of the peptide – i.e., the probability of the peptide to be locally unfolded and thus accessible for enzyme binding – is also important for TPST sulfation (Fig. 4A). We propose a simple energy cost function for local unfolding of the peptide,  $E_{\alpha}P_{\alpha} + E_{\beta}P_{\beta} - E_{rsa}P_{rsa}$ , where the coefficients  $E_{\alpha}$  and  $E_{\beta}$  are the energy costs for the unfolding of  $\alpha$ -helix and  $\beta$ -sheet, respectively. The coefficient  $E_{rsa}$  corresponds to the energy cost for losing tertiary contacts. Thus, the total effective energy for TPST sulfation can be approximated as  $E^{eff} = E_{\alpha}P_{\alpha} + E_{\beta}P_{\beta} - E_{rsa}P_{rsa} + \Delta\Delta G_{bind} + C$ , where  $C$  is an arbitrary reference coefficient.



**Figure 4. Effective sulfation energy.** (A) A schematic of tyrosine sulfation where the tyrosine-containing peptide is structured and/or buried. In order for the sulfation to occur, substrate protein undergoes a local unfolding around the sulfated tyrosine. (B) The probability density of the effective binding energy  $E^{eff}$  for sulfated and nonsulfated sequences.  $E^{eff}$  separates the two sets of sequences more effectively, compared to  $\Delta\Delta G_{bind}$  (Fig. 3B). The reference coefficient  $C$  of  $E^{eff}$  is chosen as -14.6 kcal/mol so that the two distributions intersect around zero. The Z-score between the two distributions is -1.03.

We parameterize the coefficients  $E_{\alpha}$ ,  $E_{\beta}$ , and  $E_{rsa}$  by minimizing the Z-score of  $E^{eff}$  values between the sulfated and nonsulfated datasets (see Materials and Methods),  $E_{\alpha} = 19.1$  kcal/mol,  $E_{\beta} = 24.7$  kcal/mol, and  $E_{rsa} = 7.5$  kcal/mol. We notice that these energy coefficients are physically sound. For instance, if all the

nine residues of the peptide are forming a helix, the average energy cost for breaking a backbone hydrogen bond upon unfolding would be about 2 kcal/mol, which is close to the estimated energy of a single hydrogen bond (Deechongkit *et al.*, 2004). The distributions of  $E^{eff}$  for sulfated and nonsulfated datasets are shown in Fig. 4B. Compared to  $\Delta\Delta G_{bind}$ , the separation between the two datasets with the effective sulfation energy is significantly improved with the Z-score decreased from -0.83 to -1.03. Therefore, with the effective sulfation energy  $E^{eff}$  the distributions of the sulfated and nonsulfated sequences are now separated by approximately one standard deviation.

Our results suggest that both the substrate peptide's structural properties in the host protein and its binding affinity to the enzyme are both important for the recognition by TPST. One interesting question is which term in the effective energy contributes the most to differentiating the two datasets. To answer this, we compute the Z-score for each term separately. The values for the peptide-binding energy  $\Delta\Delta G_{bind}$ , propensities of  $\alpha$ -helix  $P_{\alpha}$ ,  $\beta$ -sheet  $P_{\beta}$ , and relative solvent accessibility  $P_{rsa}$  in the host proteins, are -0.83, -0.23, -0.29, and 0.42, respectively. Therefore, the peptide-binding affinity plays a major role for TPST specificity while the thermodynamic availability of the peptide in its host protein also plays a significant role. These two factors together determine the TPST sulfation specificity, resulting into great variations in both sequences and structures of the sulfated proteins.

A better separation between sulfated and nonsulfated sequences with  $E^{eff}$  than other parameters allows us to use  $E^{eff}$  as the predictor to estimate whether a peptide is potentially sulfated. The  $E^{eff}$  distributions of sulfated and nonsulfated proteins are both Gaussian-like (Fig. 4B). We choose a cutoff value  $E_c$  corresponding to the intersection of these two distributions such that a protein has a high probability of being sulfated than nonsulfated if  $E^{eff} < E_c$ . We set the reference coefficient  $C$  of  $E^{eff}$  as -14.6 kcal/mol so that  $E_c = 0$ . As a result, the error rates for both sulfated (i.e., sulfated proteins with  $E^{eff} > 0$ ) and nonsulfated (i.e., nonsulfated proteins with  $E^{eff} < 0$ ) equal to ~15%. Using the same datasets, we compare the performance of our method with respect to two existing sulfation prediction servers, Sulfinator (Monigatti *et al.*, 2002) and PredSulSite (Huang *et al.*, 2012). Sulfinator successfully predicts 126 out of 160 sulfated and 155 out of 159 nonsulfated sites, resulting in error rates of ~21% and ~2.5%, respectively. PredSulSite succeeds in predicting 94 out of 160 sulfated and 158 out of 159 nonsulfated sites, and thus, the corresponding error rates are ~41% and ~0.6%. Interestingly, both Sulfinator and PredSulSite have a high success rate for nonsulfated sequences but with a sacrifice of the success rate for sulfated sequences. On average, our structure-based method has similar prediction accuracy as these statistics-based tools with a better prediction of the sulfated sequences. Next, we examine the applications of effective sulfation energy  $E^{eff}$  in two case studies that are not included in the datasets.

**Sulfated tyrosines in HIV-1 antibodies.** Two human monoclonal HIV antibodies, 412d and 47e, have been identified and purified from patients. The antibodies contain sulfated tyrosines in their variable loops (i.e., CDR3) that compete with the sulfated CCR5 receptor for the glycoprotein gp120 of HIV-1. It has been shown that tyrosine-sulfation is critical for both antibody binding and also for the efficiency of the viral infection (Tyrosine Sulfation of Human Antibodies Contributes to Recognition of the CCR5 Binding Region of HIV-1 gp120, 2003; Farzan *et al.*, 1999). Antibody 412d

has four tyrosines (Y96, Y100, Y100c, and Y100L, Table 1) in the CDR3 loop, and antibody 47e has three tyrosines (Y100a, Y100g and Y100h) in the loop. The question is which tyrosine or tyrosines are sulfated by TPST.

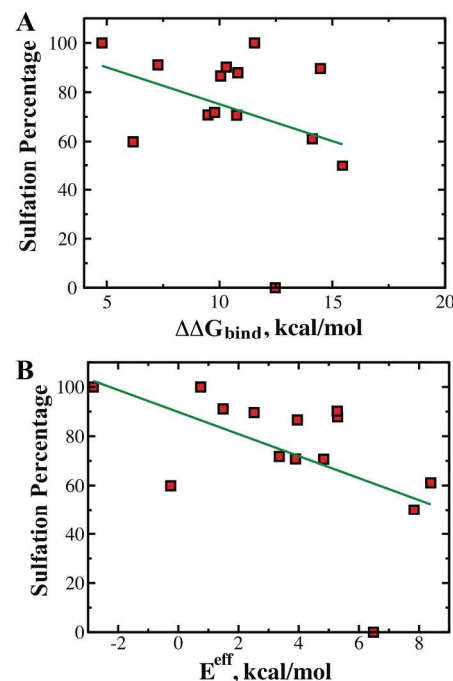
We compute for each tyrosine the corresponding 9-residue peptide's relative binding affinities,  $\Delta\Delta G_{bind}$ , and also the effective sulfation energy  $E^{eff}$  (Table 1). Since all tyrosines are positioned in the variable loop, the correction of  $E^{eff}$  compared to  $\Delta\Delta G_{bind}$  is relatively small. For antibody 47e, Y100a has the lowest  $\Delta\Delta G_{bind}$  among the three tyrosines and corresponds to the only one with  $E^{eff} < 0$ , suggesting Y100a been sulfated by TPST. The corresponding values are also consistent with other sulfated sequences, suggesting that Y100a is sulfated (Figs. 2B & 4B). The result is consistent with mutagenesis experiments (Choe *et al.*, 2003). For antibody 412d, we find that Y100c has significantly lower values of  $\Delta\Delta G_{bind}$  and  $E^{eff}$  than other three tyrosines, and the corresponding negative  $E^{eff}$  value suggest that it is sulfated. However, mutagenesis experiment found that Y100 is also sulfated in addition to Y100c. The unfavorable  $\Delta\Delta G_{bind}$  and  $E^{eff}$  values of Y100 in our calculations are caused by its upstream proline residue P97, which has strong preference for the backbone dihedral angles. The particular proline residue is not compatible with the backbone conformation of the TPST-ligand complex, resulting in an unfavorable conformation. The artifact can be resolved by modeling the backbone conformational flexibility, whose accurate and rapid characterization is still computationally challenging and is the subject for the future studies. Taking together, our method can be used to predict the potential sulfation sites although there is room for improvement.

**Table 1.** The relative binding energy  $\Delta\Delta G_{bind}$  and the effective sulfation energy  $E^{eff}$  are computed for different tyrosines of interest for HIV antibodies, 412d and 47e. The unit is kcal/mol. The contributions of secondary structure unfolding and solvent exposure are also shown. The sequence of the variable loop, CDR3, in the antibody 47e is GGEDGDYLSDPFY<sup>Y</sup>NHGMDVW, where the examined tyrosines are 100a, 100g and 100h, shown in bold face and underlined. The CDR3 sequence of the antibody 412d is YCASPY<sup>P</sup>NDY<sup>Y</sup>NDY<sup>Y</sup>AEPEGMSWY<sup>Y</sup>FDL, where the examined tyrosines are 96, 100, 100C and 100L. The experimentally validated tyrosines that undergo sulfation are colored blue.

Antibody	Index	$\Delta\Delta G_{bind}$	$E_{\alpha}P_{\alpha}$	$E_{\beta}P_{\beta}$	$-E_{rsa}P_{rsa}$	$E^{eff}$
47e	<b>Y100a</b>	-1.51	0	0	-2.94	-19.05
	<b>Y100g</b>	24.54	0	0	-2.61	7.33
	<b>Y100h</b>	22.21	0	0	-2.39	5.22
412d	Y96	31.60	0	10.99	-1.76	26.23
	<b>Y100</b>	25.59	0	0	-2.83	8.16
	<b>Y100c</b>	7.03	0	0	-3.01	-10.58
	Y100L	28.62	0	2.75	-1.65	15.12

**Sulfation efficiency.** Incomplete sulfation is often observed for many sulfated proteins. For example, gastrin, a regulator of gastric acid secretion for digestion, is partially sulfated (Bundgaard *et al.*, 1997). Systematic mutagenesis studies have been applied in order to understand the sequence dependence of the sulfation efficiency, where the extent of sulfation was measured for many mutants with mutations around the sulfated tyrosine (Bundgaard *et al.*, 1997). The extent of sulfation measurement is more quantitative than the all-or-none descriptions of the sulfated and nonsulfated datasets. In addition to the descriptive insight of the sequence dependence of sulfation efficiency derived from the previous mutational studies, a more quantitative analysis is necessary. We hypothesize that the

extent of sulfation should depend on both the thermodynamic availability of the peptide as well as the peptide-binding affinity, which are described by the effective sulfation energy,  $E^{eff}$ . We postulate that the sulfation efficiency should monotonically decrease with respect to the sulfation energy. For wild type and mutant gastrin (see Table S1 for the list of mutations), we compute the  $\Delta\Delta G_{bind}$  and  $E^{eff}$  and presented the scatter plot with respect to the sulfation efficiency in Fig. 5. Although many mutations are single-amino acid substitutions, some mutations induce changes in secondary structure propensities, and thus significantly affect the energy cost for local unfolding with respect to the host protein. As a result, we observe significant difference between  $\Delta\Delta G_{bind}$  and  $E^{eff}$  in terms of correlation with respect to sulfation efficiency. As expected, we find a better correlation between the sulfation efficiency and  $E^{eff}$  than that of  $\Delta\Delta G_{bind}$ . Using a linear regression, we find that absolute value of the correlation coefficient is improved from 0.31 for  $\Delta\Delta G_{bind}$  to 0.53 for  $E^{eff}$ . Due to inaccuracies in both experimental measurements and computational estimation of the effective energy, we do not expect a perfect correlation. Moreover, we also do not expect a linear correlation between the extent of sulfation and the sulfation energy for a wide range of energies, although the linear approximation can exist within a certain energy window. Therefore, the independent test suggests that the simple effective energy of sulfation can be used to predict the sulfation efficiency in incomplete sulfation.



**Figure 5. Correlation between sulfation efficiency and sulfation energy.** (A) The scatter-plot of the experimentally determined sulfation efficiency and computed relative peptide-binding affinity  $\Delta\Delta G_{bind}$  for gastrin and its 13 mutants. Two quantities show weak correlation, with a Pearson correlation coefficient of -0.31. (b) Better correlation is achieved by using effective sulfation energy  $E^{eff}$  in which energy costs for unfolding and desolvation are incorporated. The correlation coefficient is improved to -0.53.

**TPST-1 isoform.** Our estimation of relative peptide-binding affinity is based on the structure of TPST-2. The question is whether TPST-1 and TPST-2 have different binding affinities with respect to a given peptide. Since the sequence identity between two isoforms is as high as 64% (sequence similarity is ~78%), we ex-



pect their structures are highly similar to each other. We align the two sequences and build the homology model for TPST-1 using TPST-2 as the template (Fig. S1). We find that the peptide-binding pocket is almost identical for both isoforms with only two mutations near the N-terminal of the peptide (Fig. S1B). One of the mutations is arginine to lysine, which maintains the charge. Therefore, we postulate that a peptide would have similar binding affinities with respect to TPST-1 and TPST-2. To test this hypothesis, we calculated the relative binding affinities  $\Delta\Delta G_{bind}$  of HIV-1 antibodies and gastrin mutants to TPST-1 using the homology model. The  $\Delta\Delta G_{bind}$  values of TPST-1 and TPST-2 highly correlate with each other with the Pearson correlation coefficient  $r = 0.98$  (see the scatter plot in Fig. S2). The slope of a linear fit is 1.07 with an offset of 0.28, suggesting that peptides indeed have similar binding affinities to the two isoforms. Since other terms in the effective sulfation energy depend only on the structure of a substrate protein instead of the receptor, we expect that the effective sulfation energies  $E^{eff}$  with respect to the two isoforms are also similar, and thus no significant differences in terms of binding specificity and the sulfation specificity are observed.

In summary, using structure-based molecular modeling approaches, we have identified both structural and energetic determinants for TPST sulfation specificity. Our results suggest that both the thermodynamic availability of the peptide in a host protein and its binding affinity to the enzyme is important for TPST sulfation specificity. The thermodynamics availability of a peptide is determined by the energy cost of peptide local unfolding, where the peptide loses both tertiary and secondary interactions that stabilize its native structure in the host protein. The binding of the peptide by TPST is determined by its interactions with the residues in the binding pockets. The interplay of both peptide thermodynamic availability and enzyme binding affinity in determining the TPST sulfation specificity leads to the great variety in sulfated sequences and structures. We have developed an effective sulfation energy function that combines both the energy cost for peptide local unfolding and the binding affinity. Case studies suggest that the simple sulfation energy function can be used to predict the potential sulfation sites and the sulfation efficiency for incomplete sulfation. The benchmark study indicates that the predictive power of our structure-based sulfation predictor is comparable to other statistics-based tools with a better positive sulfation predication rate. Beside the dependence of structural availability, the major differences between structure-based and statistics-based methods include computational efficiency and transferability. Owing to a large number of calculations of inter-atomic interactions in our structure-based method, the statistics-based methods are often computationally more efficient. However, since our method is based on physical interaction, we expect our method to be transferable, such as estimating the effect of mutations in the TPST enzyme on sulfation specificity and sulfation prediction for other PTM TPST variants, such as the recently discovered bacterial TPSTs (Han *et al.*, 2012). We also expect that the proposed sulfation mechanism is also applicable to other post-translational modifications systems where the sequence or structural specificities are not well defined.

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## Supporting Information

### Structural and Energetic Determinants of Tyrosylprotein Sulfotransferase Sulfation Specificity

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**Table S1.**

The sequences of wild type and mutants of gastrin used to study the correlation between sulfation rate and sulfation energy [1]. The mutated residues are highlighted in bold.

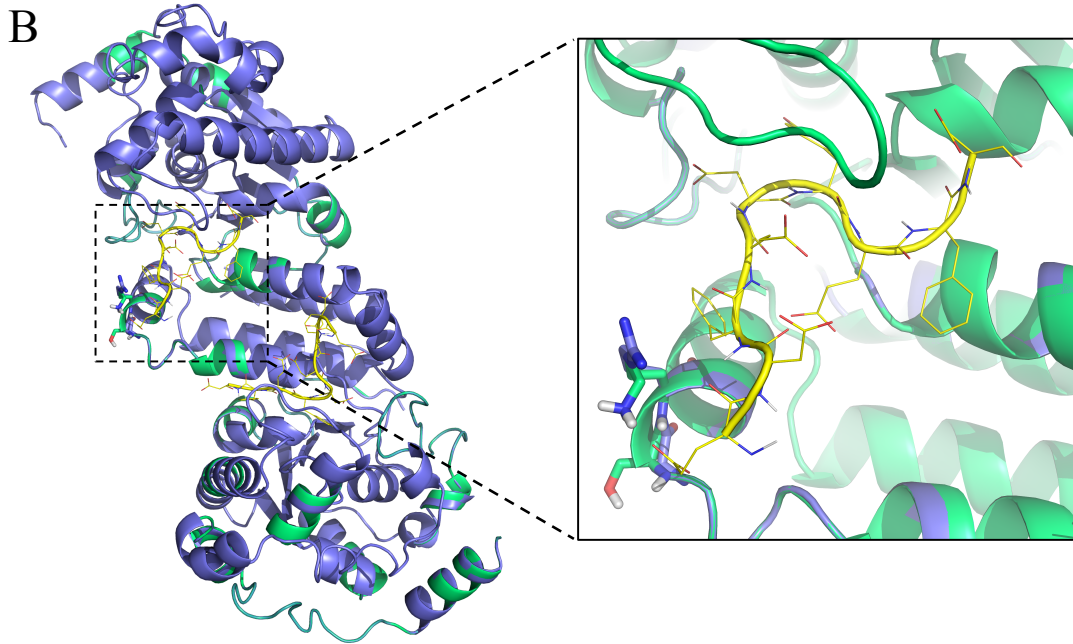
Peptide	Sequence	Sulfation Rate (%)	Sulfation Energy (kcal/mol)
Wild type gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu-Glu-Glu-Glu-Ala-Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	71.8 ± 7.2	3.36
YE-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu-Glu-Glu-Glu-Ala-Tyr- <b>Glu</b> -Trp-Met-Asp-Phe-NH <sub>2</sub>	59.9 ± 1.5	-0.26
YA-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu-Glu-Glu-Glu-Ala-Tyr- <b>Ala</b> -Trp-Met-Asp-Phe-NH <sub>2</sub>	90.2 ± 1.7	5.28
EDY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu-Glu-Glu-Glu- <b>Asp</b> -Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	100.0 ± 0.0	-2.83
ERY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu-Glu-Glu-Glu- <b>Arg</b> -Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	0.0 ± 0.0	6.48
KAY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu-Glu-Glu-Glu- <b>Lys</b> -Ala-Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	86.6 ± 3.9	3.96
AAY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu-Glu-Glu-Glu- <b>Ala</b> -Ala-Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	91.1 ± 3.1	1.49
AEAY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu-Glu- <b>Ala</b> -Glu-Ala-Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	61.1 ± 1.7	8.39
AEEAY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu- <b>Ala</b> -Glu-Glu-Ala-Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	70.8 ± 1.4	3.89
AEEEEAY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu- <b>Ala</b> -Glu-Glu-Glu-Ala-Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	70.7 ± 6.2	4.83
HDHDY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu- <b>His-Asp-His-Asp</b> -Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	100.0 ± 0	0.74
HDHAY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu-Glu- <b>His-Asp-His</b> -Ala-Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	89.6 ± 1.1	2.52
KEEAAY-gastrin	pGlu-Gly-Pro-Trp-Leu-Glu- <b>Lys</b> -Glu-Glu- <b>Ala</b> -Ala-Tyr-Gly-Trp-Met-Asp-Phe-NH <sub>2</sub>	87.9 ± 2.4	5.29

#### Reference:

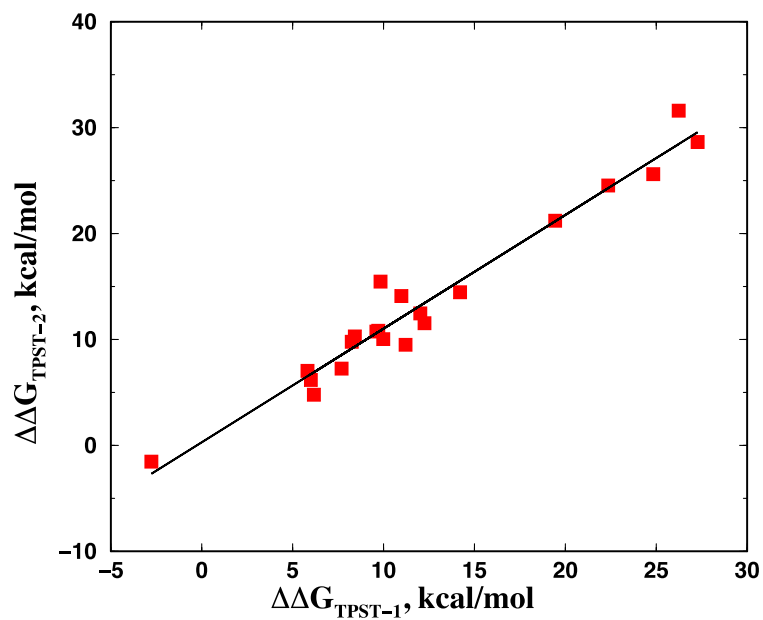
[1] J. R. Bundgaard, J. Vuust, and J. F. Rehfeld, “New Consensus Features for Tyrosine O-Sulfation Determined by Mutational Analysis,” *J. Biol. Chem.*, vol. 272, no. 35, pp. 21700–21705, Aug. 1997.

**A**

TPST-1	5	LKQNLLACLVISSVTVFYLGQHAMECHHRIEERSQPVKLESTRTTVRTGLD----	LKAN	60
TPST-2	5	+++ LL A + V LGQ +EC R+ L S R +R + + N		
TPST-1	58	VRRVLLAAGCALVLVLAVQLGQQVLEC-----RAVLAGLRSPRGAMRPEQEELVMVGTN		58
TPST-1	61	KT-FAYHKDMPILIFIGGVPRSGTTLMRAMLDAHPDIRCGEETRVIPRILALKQMWRSR	SK	119
TPST-2	59	+ Y K MPLIF+GGVPRSGTTLMRAMLDAHP++RCGEETR+IPR+LA++Q WS+S +	GR	118
TPST-1	120	HVEYRYGKAMPLIFVGGVPRSGTTLMRAMLDHPEVRCGEETRIIPRVLAMRQAWSKS		118
TPST-1	120	EKIRLDEAGVTDEVLD+AMQAFLEIIVKHGEPAPYLCNKDPFALKSLTYLSRLFPNAKF		179
TPST-2	119	EK+RLDEAGVTDEVLD+AMQAF+LE+I KHGEPA LCNKDPF LKS YLSRLFPN+KF		178
TPST-1	180	EKLRLDEAGVTDEVLDAAQAFILEVIAXKHGEPARVLCNKDPFTLKS		178
TPST-1	180	LLMVRDGRASVHSMISRKVTIAGFDLNSYRDCLTKWNRAIETMYNQCMEVGKCKMLVHY		239
TPST-2	179	LLMVRDGRASVHSMI+RKVTIAGFDL+SYRDCLTKWN+AIE MY QCMEVG +KC+ V+Y		238
TPST-1	240	LLMVRDGRASVHSMITRKVTIAGFDLSSYRDCLTKWNKAIEVMYAQCMEVGKEKCLPVYY		238
TPST-1	240	EQLVLHPERWMRTLLKFLQIPWNHSLVHHEMIGKAGGVSLSKVERSTDQVIKPVNVGAL		299
TPST-2	239	EQLVLHP R ++ +L FL I W+ +VLHHE++IGK GGVSLSK+ERSTDQVIKPVN+ AL		298
TPST-1	300	EQLVLHPRRSLKLILDFLGIAWSDAVLHHEDLIGKPGGVSLSKIERSTDQVIKPVNLEAL		298
TPST-1	300	SKWVGKIPDPVLQDMAVIAPMLAKLGYDPYANPPNYGKPDPKIIENTRRVYKGEFQLPDF		359
TPST-2	299	SKW G IP DV++DMA IAPMLA+LGYDPYANPPNYG PDP +I NT+RV KG+++ P		358
TPST-1	360	SKWTGHIPGDVVRDMAQIAPMLAQLGYDPYANPPNYGNPDPFVINNTQTVLKGDIKTPAN		358
TPST-1	360	LKEKPQTEQ	368	
TPST-2	359	LK Q Q		
TPST-2	359	LKGYFQVNQ	367	



**Figure S1. Comparison between two isoforms of TPST.** (A) The sequence alignment of two TPST isoforms, TPST-1 and TPST-2. The highlighted region indicates the two mutations near the binding pockets. One of the mutations maintains the charge. The difference between the TPST-1 and TPST-2, with respect to the peptide-binding is thus not significant. (B) The crystal structure of TPST-2 (purple) and a homology model of TPST-1 (green) are very similar. Thus the method described in this work should be transferable to TPST-1.



**Figure S2. Sulfation energies estimated using two isoforms of TPST.** The scatter plot of  $\Delta\Delta G_{\text{bind}}$  values for TPST-1 and TPST-2 features a high correlation with the Pearson correlation coefficient  $r = 0.98$ . A linear regression results in the slope of 1.07 and an offset of 0.28 kcal/mol.